

## Year 12 – Applied Science

<b>Topics</b>	Exam preparation	Learning aim C – unit 9
<b>Prior knowledge / skills</b>	Students will prepare for their Unit 1 exam by consolidating their learning and applying knowledge to exam questions.	Female and male reproductive system
<b>Key concepts / knowledge / skills covered this half term</b>		<ul style="list-style-type: none"> <li>● Structure and function of reproductive anatomy</li> <li>● Reproductive process</li> <li>● Stages in the interactions of hormones</li> <li>● Gamete development and release</li> <li>● Hormonal changes in the menstrual cycle</li> <li>● Processes leading to conception</li> <li>● Contraceptive methods</li> </ul>
<b>Assessment</b>	UCAS exams will be in June	
<b>Personal Development opportunities</b>	Students understand the organisation and function of waves in the world around us and the maintenance of homeostatic mechanisms in the human body.	
<b>Homework requirements</b>	Homework will be set on Google Classroom every week and will take 5 hours. Students should undertake individual independent study tasks by completing exam practice equations and making notes from their specification. Students should also be working on their coursework for Unit 9 independently to meet the requirements for a pass, merit or distinction.	

## Year 12 – Biology

<b>Topics</b>	Exam questions and walkthroughs
<b>Prior knowledge / skills</b>	Students have completed the content for Modules 1-4. To consolidate their learning, students will be sitting regular topic tests in lessons as well as going through exam questions to prepare for their assessments in June.
<b>Key concepts / knowledge / skills covered this half term</b>	
<b>Assessment</b>	UCAS exams will take place at the end of June
<b>Personal Development opportunities</b>	Scientific investigations require thorough practice and planning. Students will conduct practicals and follow them up with write-ups in a scientific manner.
<b>Homework requirements</b>	Homework will be set on Google Classroom every week and will take 5 hours. Students should undertake individual independent study tasks by completing exam practice questions and making notes from their specification.

## Year 12 – Chemistry

Topics	Organic synthesis	Analytical techniques	Alcohols
<b>Prior knowledge / skills</b>	<ul style="list-style-type: none"> <li>Rates of reactions and activation energy</li> </ul>	<ul style="list-style-type: none"> <li>Molar masses</li> </ul>	<ul style="list-style-type: none"> <li>Functional group of alcohols.</li> <li>Primary, secondary and tertiary carbon structures.</li> </ul>
<b>Key concepts / knowledge / skills covered this half term</b>	<ul style="list-style-type: none"> <li>Techniques and procedures used for the preparation and purification of organic solids involving use of a range of techniques</li> <li>Organic preparation; use of Quickfit apparatus; distillation &amp; heating under reflux (ii) purification of an organic solid; filtration under reduced pressure; recrystallisation; measurement of melting points</li> <li>Learn synthetic routes for several functional groups</li> </ul>	<ul style="list-style-type: none"> <li>Use of an infrared spectrum of an organic compound to identify: (i) an alcohol from an absorption peak of the O–H bond; (ii) an aldehyde or ketone from an absorption peak of the C=O bond; (iii) a carboxylic acid from an absorption peak of the C=O bond and a broad absorption peak of the O–H bond</li> <li>Interpretations &amp; predictions of an infrared spectrum of familiar or unfamiliar substances using supplied data</li> <li>Use of a mass spectrum of an organic compound to identify the molecular ion peak and hence to determine molecular mass</li> <li>Analysis of fragmentation peaks in a mass spectrum to identify parts of structures</li> </ul>	<ul style="list-style-type: none"> <li>Properties of alcohols – polarity of alcohols &amp; an explanation of water solubility in terms of hydrogen bonding</li> <li>Classification of alcohols into primary, secondary and tertiary</li> <li>Reactions of alcohols: combustion and oxidation</li> <li>Elimination of water from alcohols in the presence of an acid catalyst</li> <li>Substitution with halide ions in presence of acid to form haloalkanes</li> </ul>
<b>Assessment</b>	UCAS exams in June & students will be assessed regularly in-class with 20 minute tests.		
<b>Personal Development opportunities</b>	Industrial uses of chemistry concepts to maximise yield whilst also considering costs and safety of workers. Benefits to the environment of improved sustainability weighed against toxicity of some catalysts.		
<b>Homework requirements</b>	Homework will be set on Google Classroom every week and will take 5 hours. Students should undertake individual independent study tasks by completing exam practice equations and making notes from their specification.		

## Year 13 – Applied Science (Unit 2)

<b>Topics</b>	Unit 2
<b>Prior knowledge / skills</b>	Students will be working through and amending their Unit 2 coursework now that they have completed their exams for Unit 3. Students who have opted for a resit of previous exam units will need to spend independent time preparing for these exams.
<b>Key concepts / knowledge / skills covered this half term</b>	
<b>Assessment</b>	N/A
<b>Personal Development opportunities</b>	Scientific investigations require thorough practice and planning. Students will conduct practicals and follow them up with write-ups in a scientific manner.
<b>Homework requirements</b>	Homework will be set on Google Classroom every week and will take 5 hours. Students should undertake individual independent study tasks by completing exam practice equations and making notes from their specification.

## Year 13 – Biology

<b>Topics</b>	Exam questions and walkthroughs
<b>Prior knowledge / skills</b>	Students have completed the content for Modules 1-6. To consolidate their learning, students will be sitting regular topic tests in lessons as well as going through exam questions to prepare for their assessments in June.
<b>Key concepts / knowledge / skills covered this half term</b>	
<b>Assessment</b>	Paper 1: 5th June Paper 2: 14th June Paper 3: 19th June
<b>Personal Development opportunities</b>	Scientific investigations require thorough practice and planning. Students will conduct practicals and follow them up with write-ups in a scientific manner.
<b>Homework requirements</b>	Homework will be set on Google Classroom every week and will take 5 hours. Students should undertake individual independent study tasks by completing exam practice equations and making notes from their specification.

## Year 13 – Chemistry

<b>Topics</b>	Lattice enthalpy	Exam questions and walkthroughs
<b>Prior knowledge / skills</b>	<ul style="list-style-type: none"> <li>● Enthalpy cycles</li> <li>● Ionic compounds</li> </ul>	Students have completed the content for Modules 1-6. To consolidate their learning, students will be sitting regular topic tests in lesson as well as going through exam questions to prepare for their assessments in June
<b>Key concepts / knowledge / skills covered this half term</b>	<ul style="list-style-type: none"> <li>● Explanation of the term lattice enthalpy</li> <li>● Born-Haber and related enthalpy cycles</li> <li>● Use of lattice enthalpy of simple ionic solid &amp; relevant energy terms</li> <li>● Explanation &amp; use of the terms enthalpy change of solution and enthalpy change of hydration</li> <li>● Use of enthalpy change of collusion of simple ionic solids &amp; construction of enthalpy cycles with related calculations</li> <li>● Qualitative explanation of the effect of ionic charge &amp; ionic radius on the exothermic value of lattice enthalpy &amp; enthalpy change of hydration</li> </ul>	
<b>Assessment</b>	Paper 1: 10th June Paper 2: 18th June Paper 3: 21st June	
<b>Personal Development opportunities</b>	The instrumentation methods of analysis studied during the A-Level course provide learners with an important base of knowledge, understanding and awareness for further study in Higher Education and in many areas of employment in the broad scientific field.	
<b>Homework requirements</b>	Homework will be set on Google Classroom every week and will take 5 hours. Students should undertake individual independent study tasks by completing exam practice equations and making notes from their specification.	